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## Tetrabutylammonium hydroxide, 40 wt.% (1.5M) solution in water

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General		
Product Name	Tetrabutylammonium hydroxide	
CAS RN	2052-49-5	
ACD Code	MFCD00009425	
Structure		
Molecular Formula	C <sub>16</sub> H <sub>37</sub> N O	
Molecular weight	259.48	
Pack size	{ Error }	
Physical		
Density (g/cm <sup>3</sup> )	0.995	
Boiling Point (°C)	>100	
Safety		
GHS Pictogram		
GHS Signal Word	Danger	
GHS H statement	H314: Causes severe skin burns and eye damage	
GHS P statement	P280: Wear protective gloves/protective clothing/eye protection/face protection P301 + P330 + P331: IF SWALLOWED: rinse mouth. Do NOT induce vomiting P301: IF SWALLOWED P305 + P351 + P338: IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing P303 + P361 + P353: IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water/shower	
Hazard	C: Corrosive	
Risk	35: Causes severe burns.	
Safety	26: In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. 36/37/39: Wear suitable protective clothing, gloves and eye/face protection. 45: In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible).	
Categories		
Preparation, Purification and Analysis > Routine Reagents Functional Reagents > Organic Bases		
Applications		
Function Transformation Type	Caveat	Reference
<a href="#">Reagent</a> <a href="#">Organic Base</a>	Alkylation and condensation reactions.	SC 1983, 13, 927
Other		
Parameter	EINECS	218-147-6
	Solubility	Solubility in water: soluble
	Origin	synthetic
	References:	Basic nonaqueous titrant. Anal. Chem., 34, 584 (1962). Reagent for ion-pair extractions. Anions from weak acids can be extracted with CHCl <sub>3</sub> or CH <sub>2</sub> Cl <sub>2</sub> . Acta Chem. Scand., 23, 2202, 2203, 2204 (1969).
	Literature Reference:	04,11,634
	Beilstein Reference:	05,645; 11,500
	Fieser	
3D model	<a href="#">Show</a>	